CASE GT/3-21923/A/AC 533/CONT

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

IN RE APPLICATION OF

Group Art Unit: 1711

ANNE FLISHER ET AL.

Examiner: S. Berman

CONTINUATION OF APPLICATION NO: 09/890,129

FILED CONCURRENTLY HEREWITH

FOR: POLYMERISATION PROCESS

Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313-1450

INFORMATION DISCLOSURE STATEMENT

Sir:

In accordance with 37 CFR 1.56, Applicants wish to call the Examiner's attention to the references cited on the attached form PTO-1449.

DE 4123889 and English language abstract are enclosed herewith.

The Examiner is requested to consider the foregoing information in relation to this application and indicate that each reference was considered by returning a copy of the initialed PTO 1449 form.

Respectfully submitted,

Ciba Specialty Chemicals Corporation Patent Department

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Encl. References

PTO-1449 Form

Tyler A. Stevenson Agent for Applicants Reg. No. 46,388

Sheet <u>1</u> of <u>1</u>.

FORM PTO-1449				Docket Number (Optional)		Application	Application Number		
INFORMATION DISCLOSURE OTATION				GT/3-21923/A/AC 5					
INFORMATION DISCLOSURE CITATION IN AN APPLICATION (Use several sheets if necessary)				Applicant ANNE FLISHER ET AL.					
				Filing Date	Group Art	Group Art Unit			
				March 2, 2004					
			U. S. PATEN	T DOCUMENTS					
EXAMINER									
INITIAL	DOCUMENT NUMBER	DATE	NAME		CLASS	SUBCLASS	APPROPRIATE		
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	<u>I</u>	FC	REIGN PATI	ENT DOCUMENTS					
							Translation		
	DOCUMENT NUMBER	DATE	,	COUNTRY	CLASS	SUBCLASS	YES	NO	
	4123889	3/92	Germany						
	OTHE	R DOCU	MENTS (includi	ng Author, Title, Date, Pertin	ent Pages, Etc	.)			
	Derwent Abstract 92-089567/12 for DE 4123889 (1992)								
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EXAMINER	<u> </u>			DATE CONCIDENCE					
				DATE CONSIDERED					
EXAMINER: Init	ial if citation considered.	whether or no	ot citation is in conf	formance with MPEP §609; I	Oraw line throu	gh	-0		
				m with next communication					

92-089567/12

A35 D22 (A18 A25 A93 A96)

SANN 07.09.90

A(7-A, 10-E10, 10-G1A, 12-V3A) D(9-C4B)

SANYO CHEM IND LTD

07.09.90-JP-238514 (12.03.92) C08f-02/38 C08f-06 C08f-251 C08f-291 C08j-03/28

Water-absorbing resin prepn. - by irradiating polymer of water-soluble menomer and polysaccharide and/or crosslinking agent with UV in presence of radical scavenger C92-041254

The prepn. of water-absorbing resins (i) with reduced content of residual moremer and water-soluble components comprises irradiating (1), obtd. by polymsg. a water-soluble monomer with a polysaccharide and/or crosslinking agent, with UV-radiation in the presence of a radical scavenger in every drying or pulverising stage after polymsn..

Also claimed are (I) comprising 500 ppm or less residual monomer and 7 wt. 4 or less water-soluble components.

USE/ADVANTAGE

(i) contain reduced amts. of residual monomer and water soluble components. (I) are used in prods. in contact with the human body, e.g. fuid-absorbing pads for sanitary prods. and bandages.

MORE SPECIFICALLY

The amt. of radical scavengers used is 0.001-5% of the total wt. of polymerisable monomer and crosslinking agent.

EXAMPLE

196g Acrylic acid. 0.05g methylenebisacrylamide and 236g deionised water were mixed together and 168g aq. soln. contg. 48% NaOH was added gradually, keeping the temp. under 50°C. to neutralise approx. 74 mol.% of the acrylic acid. The concn. of the dissolved oxygen was reduced to 1 ppm or less by adding nitrogen. 0.05g V-50 (RTM; azotype polymen. initiator) was added to the soln and mixed for limb. for 1min.

The resulting soln. was poured into a steel container contg. exygen sealed with polyethane film and polymsd. for 1 hr. in a water bath at 50°C to produce a hydrogel polymen An sq. soln. (prepd. by dissolving 0.72g hydroquinone in 14g water) was sprayed evenly over the surface 'f 600g of

the polymer.

The gel was placed on a conveyor belt and irradiated for 10 sec. with UV-radiation (80 W/cm). The gel was grawked and dried at 130°C in air. The dried nolymer was reduced

to less than 20 mesh.
The residual monomer content was 230 ppm, the water soluble component confent was 3.6% and the water absorption was 62 g/g. (9pp2223MODwgNo0/0). I DE4123889- 3